

Download Chapter 5 Electrons In Atoms Answer Key

Chapter 2: Atoms and the Periodic Table This content can also be downloaded as a printable PDF, Adobe Reader is required for full functionality. This text is published under Creative Commons licensing, for referencing and adaptation, please click here. 2.1 Atomic Theory with Historical Perspectives 2.2 Introduction to Elements and the Periodic ...Key Concepts . The attractions between the protons and electrons of atoms can cause an electron to move completely from one atom to the other. When an atom loses or gains an electron, it is called an ion. Key Concepts. Atoms are made of extremely tiny particles called protons, neutrons, and electrons. Protons and neutrons are in the center of the atom, making up the nucleus. In ordinary chemical reactions, the nucleus of each atom (and thus the identity of the element) remains unchanged. Electrons, however, can be added to atoms by transfer from other atoms, lost by transfer to other atoms, or shared with other atoms.